

EUDESMANE ALCOHOLS FROM *JASONIA GLUTINOSA*

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Key Word Index — *Jasonia glutinosa*; Compositae; sesquiterpenoids; eudesmane derivatives.

Abstract—Three new sesquiterpene alcohols have been isolated from *Jasonia glutinosa*. Their structures were elucidated by spectroscopic methods and chemical correlations as (–)-[11*R*]-4 α ,14-epoxyeudesm-11,12-diol, (–)-[11*R*]-eudesm-4(14)-en-5 β ,11,12-triol and (+)-[11*R*]-eudesm-4(14)-en-5 α ,11,12-triol and they are called α -epoxy kudtdiol, 5-*epi*-kudtriol and kudtriol respectively.

INTRODUCTION

In a previous paper we reported the structural determination of kudtdiol (**1**), the major sesquiterpene component of *J. glutinosa* (family Compositae; tribe Inuleae), as (+)-[11*R*]-eudesm-4(14)-en-11,12-diol [1]. The same structure had previously been proposed to sesquibenhidiol, a substance isolated from *Chamaecyparis formosensis* [2], but the physical constants reported (mp 123°, [α]_D – 38°) differed from those found for kudtdiol (mp 90°, [α]_D + 72.9°). Recently, Bohlmann [3] isolated from another species of Compositae, *Flourensia heterolepis* (Heliantheae), an oily product ([α]_D + 50.4°) whose spectroscopic properties corresponded to those of kudtdiol. In addition other sesquiterpene acids with a eudesmane skeleton were isolated.

In this report we describe the structures of three new sesquiterpene alcohols with eudesmane and 5-*epi*-eudesmane skeletons, which we isolated from *Jasonia glutinosa*.

RESULTS AND DISCUSSION

The neutral fraction of the benzene extract from the aerial parts of *J. glutinosa*, after dry column chromatography, mostly yielded kudtdiol (**1**). From a more polar fraction of the chromatography, α -epoxykudtdiol (**2**), 5-*epi*-kudtriol (**3**) and kudtriol (**4**) were isolated.

α -Epoxy-kudtdiol (**2**)

The IR spectrum of **2** showed absorptions of hydroxyl and oxiranic groups (3400, 3020, 1260, 1040, 935, 920 cm⁻¹). The ¹H NMR spectrum had signals of a quaternary methyl (δ 0.80 s), a methyl geminal with a tertiary hydroxyl (1.06 s), an oxiranic methylene (2.48 and 2.72, AB, *J* = 5 Hz) and two geminal protons of a primary hydroxyl group (3.35 and 3.47, AB, *J* = 11 Hz). On acetylation (Ac₂O-pyridine room temp.), a hydroxyacetate **5** (IR bands at 3460, 1740 cm⁻¹) was obtained, whose ¹H NMR spectrum had the acetate methyl signal at δ 2.10 and the methylene protons geminal with the acetate group absorbing at 4.01. These data, compared with those of kudtdiol, allow us to propose structure **2**.

On the other hand, since the major product of the epoxidation of **1** with *m*-chloroperbenzoic acid has been shown to be spectroscopically identical with the natural product **2**, and since it has been established that the homoallylic axial methyl group at C-10 preferentially induces an α -attack of the epoxidizing agent [4], the configuration of the epoxy group must be that depicted in **2**. Furthermore the C-10 methyl resonances in the triol **6**, obtained on LiAlH₄ reduction of **2**, and in the triol **7**, obtained from kudtdiol acetate **8**, confirm the proposed structure.

5-*epi*-Kudtriol (**3**) and kudtriol (**4**)

Both substances had very similar chromatographic and spectroscopic properties. The less polar product **3** had M⁺ at *m/e* 254 (C₁₅H₂₆O₃) and IR absorptions for hydroxyl groups and exocyclic unsaturation. Its ¹H NMR spectrum showed signals for a quaternary methyl (δ 0.99, s), a methyl geminal with a tertiary hydroxyl (1.13, s), two geminal protons of a primary hydroxyl (3.35 and 3.55, AB, *J* = 11 Hz) and an olefinic exocyclic methylene (4.90, *br s*). Compound **3** on acetylation gave the dihydroxymonoacetate **9**, whose ¹H NMR spectrum had an acetate singlet at 2.05, and the AB methylene signals at 3.35 and 3.55 of **3** changed to a singlet at 3.99.

Substance **4** also had M⁺ at *m/e* 254 (C₁₅H₂₆O₃) and major differences in the IR and ¹H NMR spectra from those of substance **3** were the C—O absorption band at 1040 cm⁻¹ and the resonances of the C-10 methyl group (δ 0.83) and the vinylidene group (4.65 and 4.76). Compound **4** on acetylation also gave a dihydroxymonoacetate **10**, similar to **9** except for the ¹H NMR signals of the same groups.

These data suggest, for both substances, a structure like that of kudtdiol (**1**), with an additional tertiary hydroxyl located at C-5 or C-7. The observed effects on the C-10 methyl and C-14 proton resonances, compared with those of kudtdiol (**1**), allow us to propose the structure eudesm-4(14)-en-5,11,12-triol for these substances. The stereochemistry at C-5 for both triols was assigned by comparison of the relative deshielding of the C-10 methyl and the C=CH₂ resonances in **3**, which must have the C-5

hydroxyl group in the β disposition. Consequently, **4** has the opposite configuration.

The confirmation of the proposed structures **3** and **4** was achieved by partial synthesis from kudtdiol acetate (**8**), which after treatment with $H_2SO_4-HOAc-H_2O$ gave compounds **11** (33%), **12** (6%), unreacted **8** (44%), **1** (3%), **13** (5%) and **14** (9%). The saponification of **11**, followed by sensitized photooxidation [5] of **15** and $NaBH_4$ reduction led to the isolation of synthetic **3** and **4**, in agreement with the results obtained with similar substances [6]. Both compounds were identical to the natural **3** and **4** in R_f and spectral properties.

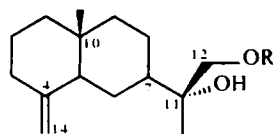
The isopropylidene ketal **16**, easily obtained from **4**, showed a pyridine-induced shift ($\Delta_{pyridine}^{CDCl_3} = -0.06$ ppm) for the C-10 methyl group which agrees with a dihedral angle of $\theta \approx 180^\circ$ with respect to the C-5 hydroxyl [7]. When ketalization of **3** was attempted a complex reaction mixture was obtained, which could be explained by possible multiple interactions between the three hydroxyl groups in the conformation **17**.

EXPERIMENTAL

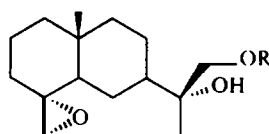
Optical rotations were measured in $CHCl_3$ soln. IR spectra in film form or $CHCl_3$ soln. 1H NMR (60 MHz) in $CDCl_3$ with int. TMS, chemical shifts are in δ , MS (70 eV) values are in *m/e* units (% relative abundance). TLC, PLC and CC separations were performed on Si gel G, Si gel $HF_{313+263}$ and Si gel 60 (Merck).

Extraction and isolation. *Jasonia glutinosa* was collected in Algora (Guadalajara, Spain) September 1976. The air-dried plant (4.897 kg) was extracted with refluxing C_6H_6 for 12 hr, yielding 151.4 g. Dewaxing with MeOH gave waxes (66.7 g) and the remaining product, on treatment with urea in MeOH (saturated soln), produced 19.7 g of linear products. The dewaxed material in Et_2O soln was extracted with 10% NaOH to yield 35.5 g of neutral fraction and 19.3 g of acid fraction. The neutral fraction by dry CC yields 17.1 g of **1** and 2.57 g of a mixture which by CC on Si gel and Ag^+-Si gel (20%) yielded 300 mg of **1**, 72 mg of **2**, 119 mg of **3** and 95 mg of **4**.

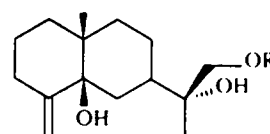
Kudtdiol (1). Mp 90° (CH_2Cl_2 -hexane) $[\alpha]_D^{25} + 72.9^\circ$ (c 2.15). IR $\nu_{max} cm^{-1}$: 3390, 3060, 1640, 1045, 880. 1H NMR: δ 0.69 (3H,



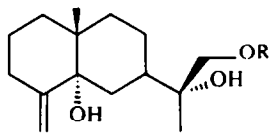
1 R = H
8 R = Ac



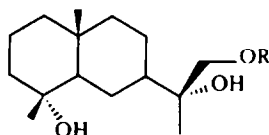
2 R = H
5 R = Ac



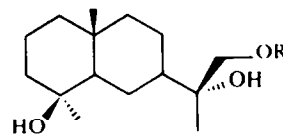
3 R = H
9 R = Ac



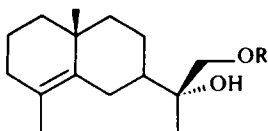
4 R = H
10 R = Ac



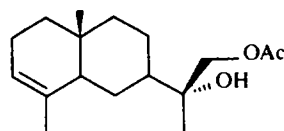
6 R = H
14 R = Ac



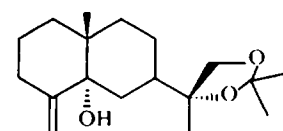
7 R = H
13 R = Ac



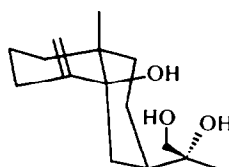
11 R = Ac
15 R = H



12



16



17

s, Me-C-10), 1.11 (3 H, s, Me-C-11), 3.38 and 3.52 (2 H, AB, $J = 11$ Hz, H-12), 4.42 (1 H, *br s*, H-14), 4.69 (1 H, *br s*, H-14). MS *m/e* (rel. int.): M^+ 238 (6), 223 (2), 220 (2), 207 (100), 189 (77), 164 (19), 149 (47).

Kudtdiol monoacetate (8). 374 mg of **1**, on acetylation (room temp.) gave 421 mg of **8**. $[\alpha]_D + 63.2^\circ$ (c 2.09). IR $\nu_{\max} \text{ cm}^{-1}$: 3460, 3060, 1740, 1640, 1235, 1040, 880. $^1\text{H NMR}$: δ 0.69 (3 H, s, Me-C-10), 1.14 (3 H, s, Me-C-11), 2.06 (3 H, s, -OAc), 4.01 (2 H, s, H-12), 4.43 (1 H, *br s*, H-14), 4.67 (1 H, *br s*, H-14). MS *m/e* (rel. int.): M^+ 280 (2), 262 (1), 220 (2), 207 (15), 202 (11), 189 (19), 163 (6), 117 (100).

α -Epoxykudtdiol (**2**). $[\alpha]_D - 6.05^\circ$ (c 0.90). IR $\nu_{\max} \text{ cm}^{-1}$: 3400, 3020, 1105, 1040. $^1\text{H NMR}$: δ 0.80 (3 H, s, Me-C-10), 1.06 (3 H, s, Me-C-11), 2.48 and 2.72 (2 H, AB, $J = 5$ Hz, H-14), 3.35 and 3.47 (2 H, AB, $J = 11$ Hz, H-12).

α -Epoxykudtdiol monoacetate (**5**). Acetylation of **2** (23 mg) gave **5** (25 mg). $[\alpha]_D - 3.20^\circ$ (c 0.62). IR $\nu_{\max} \text{ cm}^{-1}$: 3460, 3020, 1740, 1235, 1035. $^1\text{H NMR}$: δ 0.86 (3 H, s, Me-C-10), 1.14 (3 H, s, Me-C-11), 2.10 (3 H, s, -OAc), 2.50 and 2.74 (2 H, AB, $J = 5$ Hz, H-14), 3.92 and 4.10 (2 H, AB, $J = 12$ Hz, H-12).

Synthesis of α -epoxykudtdiol. Treatment of **1** (208 mg) with *m*-chloroperbenzoic acid in CH_2Cl_2 soln, followed by PLC (Et_2O) afforded α -epoxykudtdiol (94 mg) identical with the natural product **2**.

Reduction of α -epoxykudtdiol. Compound **2** (55 mg) by LiAlH_4 reduction gave (-)-[11*R*]-eudesm-4 α ,11,12-triol **6**. $[\alpha]_D - 6.9^\circ$ (c 1.01). IR $\nu_{\max} \text{ cm}^{-1}$: 3350, 1170, 1095, 1050, 905. $^1\text{H NMR}$: δ 0.86 (3 H, s, Me-C-10), 1.10 (6 H, s, Me-C-11 and Me-C-4), 3.39 and 3.53 (2 H, AB, $J = 10$ Hz, H-12). Monoacetate **14**. Mp 100° (CH_2Cl_2 -hexane). $[\alpha]_D - 4.1^\circ$ (c 2.03). IR (soln) $\nu_{\max} \text{ cm}^{-1}$: 3580, 1730, 1235, 1030, 900. $^1\text{H NMR}$: δ 0.86 (3 H, s, Me-C-10), 1.12 (3 H, s, Me-C-4), 1.17 (3 H, s, Me-C-11), 2.10 (3 H, s, -OAc), 4.02 (2 H, s, H-12).

5-epi-Kudtdiol (**3**). $\text{C}_{15}\text{H}_{26}\text{O}_3$, $[\alpha]_D - 10.4^\circ$ (c 0.79). IR (soln) $\nu_{\max} \text{ cm}^{-1}$: 3570, 3400, 3060, 1635, 1200, 1020, 900. $^1\text{H NMR}$: δ 0.99 (3 H, s, Me-C-10), 1.13 (3 H, s, Me-C-11), 3.35 and 3.55 (2 H, AB, $J = 11$ Hz, H-12), 4.90 (2 H, *br s*, H-14). MS *m/e* (rel. int.): M^+ 254 (6), 236 (18), 223 (13), 221 (25), 218 (10), 205 (74), 203 (13), 187 (65), 162 (58), 147 (100).

5-epi-Kudtdiol monoacetate (**9**). Acetylation of **3** (30 mg) gave **9** (28 mg). $[\alpha]_D - 8.2^\circ$ (c 0.37). IR $\nu_{\max} \text{ cm}^{-1}$: 3440, 3060, 1730, 1640, 1240, 1030, 900. $^1\text{H NMR}$: δ 0.99 (3 H, s, Me-C-10), 1.16 (3 H, s, Me-C-11), 2.05 (3 H, s, -OAc), 3.99 (2 H, s, H-12), 4.89 (2 H, *br s*, H-14).

Kudtdiol (**4**). $\text{C}_{15}\text{H}_{26}\text{O}_3$, $[\alpha]_D + 76.5^\circ$ (c 1.03). IR (soln) $\nu_{\max} \text{ cm}^{-1}$: 3570, 3400, 3060, 1640, 1205, 1095, 1040, 900. $^1\text{H NMR}$: δ 0.83 (3 H, s, Me-C-10), 1.06 (3 H, s, Me-C-11), 3.40 and 3.56 (2 H, AB, $J = 12$ Hz, H-12), 4.65 (1 H, *br s*, H-14), 4.76 (1 H, *br s*, H-14). MS *m/e* (rel. int.): M^+ 254 (1), 236 (2), 223 (1), 222 (2), 221 (1), 218 (1), 205 (7), 203 (2), 187 (4), 162 (10), 147 (17), 83 (100).

Kudtdiol monoacetate (**10**). Acetylation of **4** (35 mg) gave **10** (37 mg). $[\alpha]_D + 56.7^\circ$ (c 0.59). IR (soln) $\nu_{\max} \text{ cm}^{-1}$: 3580, 3420, 3060, 1730, 1640, 1240, 1040, 905. $^1\text{H NMR}$: δ 0.84 (3 H, s, Me-C-10), 1.14 (3 H, s, Me-C-11), 2.07 (3 H, s, -OAc), 4.01 (2 H, s, H-12), 4.65 (1 H, *br s*, H-14), 4.76 (1 H, *br s*, H-14).

Synthesis of 5-epi-kudtdiol (**3**) and kudtdiol (**4**). Isomerization of kudtdiol monoacetate (**8**). To HOAc soln of **8** (1.825 g in 1.86 ml) were added H_2SO_4 (1.16 ml) and H_2O (4.6 ml). After 1.75 hr, 1.820 g of the reaction mixture were recovered which on Ag^+ -Si

gel (20%_{v/v}) CC afforded several products. (+)-[11*R*]-Eudesm-4-en-11,12-diol monoacetate (**11**), (532 mg, hexane-EtOAc, 20:1) $[\alpha]_D + 97.8^\circ$ (c 1.32). $^1\text{H NMR}$ (CCl_4): δ 1.01 (3 H, s, Me-C-10), 1.10 (3 H, s, Me-C-11), 1.59 (3 H, s, Me-C-4), 2.04 (3 H, s, -OAc), 3.94 (2 H, s, H-12). The saponification of **11** gave **15**, mp 115° (hexane- CH_2Cl_2), $[\alpha]_D + 94.4^\circ$ (c 0.96). IR (soln) $\nu_{\max} \text{ cm}^{-1}$: 3540, 3400, 1090, 1030. $^1\text{H NMR}$: δ 1.01 (3 H, s, Me-C-10), 1.12 (3 H, s, Me-C-11), 1.59 (3 H, s, Me-C-4), 3.40 and 3.60 (2 H, AB, $J = 11$ Hz, H-12). (+)-[11*R*]-Eudesm-3-en-11,12-diol monoacetate (**12**), (100 mg, hexane-EtOAc, 20:1) $[\alpha]_D + 17.4^\circ$ (c 2.15). IR (soln) $\nu_{\max} \text{ cm}^{-1}$: 3570, 3010, 1735, 1650, 1230, 1035, 905, 840. $^1\text{H NMR}$: δ 0.75 (3 H, s, Me-C-10), 1.13 (3 H, s, Me-C-11), 1.58 (3 H, s, Me-C-4), 2.06 (3 H, s, -OAc), 4.30 (2 H, s, H-12); 5.30 (1 H, *m*, H-3). Kudtdiol monoacetate (**8**), (712 mg, hexane-EtOAc, 9:1). (+)-[11*R*]-Eudesm-4,11,12-triol monoacetate (**13**), (78 mg, hexane-EtOAc, 9:1), $[\alpha]_D + 8.8^\circ$ (c 2.13). IR $\nu_{\max} \text{ cm}^{-1}$: 3460, 1725, 1240, 1040. $^1\text{H NMR}$: δ 1.01 (3 H, s, Me-C-10), 1.15 (6 H, s, Me-C-11 and Me-C-4), 2.07 (3 H, s, -OAc), 4.03 (2 H, s, H-12). The saponification of **13** gave **7**. $[\alpha]_D + 10.0^\circ$ (c 0.65). IR (soln) $\nu_{\max} \text{ cm}^{-1}$: 3580, 3400, 1085, 1030. $^1\text{H NMR}$: δ 1.01 (3 H, s, Me-C-10), 1.14 (3 H, s, Me-C-11), 1.17 (3 H, s, Me-C-4), 3.43 and 3.61 (2 H, AB, $J = 10$ Hz, H-12). Kudtdiol (**1**), (49 mg, hexane-EtOAc, 1:1). (-)-[11*R*]-Eudesm-4,11,12-triol monoacetate (**6**), (143 mg, hexane-EtOAc, 1:1).

Photooxidation of (+)-[11*R*]-eudesm-4-en-11,12-diol (**15**). Compound **15** (160 mg) by oxidation (Rose Bengale as sensitizer) in *iso*-PrOH, followed by NaBH_4 reduction and CC on Ag^+ -Si gel (20%_{v/v}) afford **3** (27 mg) and **4** (36 mg).

5 α -Hydroxveudesm-4(14)-en-11,12-iso-propylidene ketal (**16**). Kudtdiol (**4**) (47 mg) was reacted with 0.8 ml of $\text{Me}_2\text{C}(\text{OMe})_2$ and TsOH in Me_2CO ; after PLC (hexane-Et₂O, 1:1) 25 mg of **16** were isolated, mp 155° (hexane- CH_2Cl_2). IR (KBr) $\nu_{\max} \text{ cm}^{-1}$: 3460, 3080, 1640, 1235, 1195, 1095, 1045, 890. $^1\text{H NMR}$: δ 0.85 (3 H, s, Me-C-10), 1.24 (3 H, s, Me-C-11), 1.35 (3 H, s, Me-C-1), 1.41 (3 H, s, Me-C-1), 3.64 and 3.92 (2 H, AB, $J = 8$ Hz, H-12), 4.69 (1 H, *br s*, H-14), 4.79 (1 H, *br s*, H-14).

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